$\begin{array}{l} \mbox{PRECURSOR-REFORMING SYNTHESIS OF MESOPOROUS} \\ \mbox{g-}C_{3}N_{4} \mbox{ FOR PHOTOCATALYTIC DESTRUCTION OF} \\ \mbox{ORGANIC DYE AND PHARMACEUTICALS} \end{array}$

Starukh H.^{1,2,3}, P. Praus ^{1, 2}

¹ Center for Energy and Environmental Technologies, VŠB-Technical University of Ostrava, 17 Listopadu 15, Ostrava, Czech Republic, halyna.starukh@vsb.cz
² Faculty of Materials Science and Technology, VŠB-Technical University of Ostrava, 17. listopadu 15,708 00 Ostrava-Poruba, Czech Republic

³Chuiko Institute of Surface Chemistry, NAS of Ukraine, 17 General Naumov Str., Kyiv 03164, Ukraine

Here we describe a novel precursor reforming strategy that ultimately affords mesoporous $g-C_3N_4$ materials with red shifted light absorption edge. The precursor synthesis was performed by hydrothermal treatment of mixtures melamine (M) with urea (U) or thiourea (TU) in aqueous or DMF medium. It was confirmed that hydrothermal treatment of melamine and urea (thiourea) causes an irreversible melamine monoclinic to orthorhombic phase transformation. The calcination of the mixture of orthorhombic melamine and urea (thiourea) yielded mesoporous $g-C_3N_4$ with enhanced photoabsorption properties.



|--|

	content, at.%				N/C atomic
	С	Ν	н	0	ratio
MUdmf-b	34.93	2.18	61.0	1.89	1.50
MUdmf-e	34.56	2.49	60.95	2.0	1.51
MTU-b	34.47	2.27	60.35	2.91	1.50
MTUdmf-b	35.38	2.19	59.87	2.56	1.45

Conclusions

- The precursor reforming strategy was implemented to synthesize the mesoporous g-C₃N₄ materials with red shifted light absorption edge.
- The hydrothermal treatment of melamine mixtures with urea causes the transformation of melamine from monoclinic to orthorhombic phase in aqueous and DMF medium.
- In the case of hydrothermal treatment of melamine and thiourea mixture in aqueous medium the precursor consist of melamine and ammonium thiocyanate. The thermal treatment this mixture in DMF medium causes the hydrazinecarbothioamide formation.
- The calcination of thiourea derivatived precursors at 550 °C. results in g-C₃N₄ with strongly red shifted light absorption edge.
- The highest activity in the dye degradation was observe for g-C₃N₄ obtained from precursor containing orthorhombic melamine.

This work was supported by the Czech Science Foundation (project No. 19-15199S), EU structural funding in Operational Program Research, Development and Education (project No. CZ.02.1.01/0.0/0.0/16_019/0000853 "IET-ER") and VŠB-TU Ostrava (project No. SP 2021/46)

Photocatalytic activity study

Cumulative

cm³q⁻¹

0.03

0.59

0.12

1.4

0.01

0.49

pore volume,

Volume

cm³g⁻

162

819

195

1781

86

597

adsorbed N₂

SSA,

cm²g[.]

49

215

30

526

180

5

MU-b

MU-e

MUdmf-b

MUdmf-e

MTUdmf-b

М-е

