

CO₂ ADSORPTION ON AMINE-FUNCTIONALIZED PERIODIC MESOPOROUS SILICA AND ORGANOSILICAS

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Nowadays, the reduction of CO₂ emission became the key issue to slow down the global warming. Recently, various adsorption techniques and materials for CO₂ capture have been developed. For decades the amine-containing materials such as liquid amines have been used in industrial processes to capture CO₂. Nevertheless, this process possesses several disadvantages, including high cost and high energy consumption associated with evaporation of CO₂. As a result, the research on the CO₂ capture is gradually shifting from liquid amine-based scrubbing to solid adsorption to overcome these drawbacks. Mesoporous silica is a very popular solid adsorbent for gas- and liquid-phase applications including CO₂ sorption. Recently, porous solid adsorbents such as silica have been modified by the amine-containing compounds to increase their affinity toward CO₂. Herein, CO₂ adsorption was investigated on amine-functionalized mesoporous silica and periodic mesoporous organosilica samples. Hexagonally ordered mesoporous SBA-15 and benzene-PMO (BPMO) samples were prepared and functionalized by amine-containing modifier TSPED (N-[3-(trimethoxysilyl)propyl]-ethylenediamine). Nitrogen adsorption isotherms showed that these samples featured mesostructured, high surface area, and narrow pore size distributions. N-[3-(trimethoxysilyl)propyl]-ethylenediamine-modified silica samples exhibited substantially higher CO₂ uptake compared to original material; maximal values for SBA-15_A ~3.1 mmol/g, for BPMO_A ~3.0 mmol/g and SBA-15-BTEP_A ~2.9 mmol/g, measured on a volumetric adsorption analyzer at 25 °C. The amine-modification also significantly improved the CO₂/N₂ separation properties of all prepared silica samples (e.g. SBA-15 from 14 to 86 for pressure 1 bar).

Acknowledgement

This work was supported from the Czech Science Foundation project no. 19-23760J.